


Name:			
Enrolment No:			
<b>UPES</b> <b>End Semester Examination, May 2024</b>			
<b>Course: Nuclear and Particle Physics</b> <b>Program: Int (BSc + MSc)</b> <b>Course Code: PHYS3035</b>		<b>Semester: VI</b> <b>Time : 03 hrs.</b> <b>Max. Marks: 100</b>	
<b>Instructions: All questions are compulsory and question no 9 and 11 have internal choices</b>			
<b>SECTION A</b> <b>(5Qx4M=20Marks)</b>			
S. No.		Marks	CO
Q 1	Briefly explain the mirror nuclei method for determining nuclear radius.	4	CO1
Q.2	The contribution of surface energy per nucleon to the binding energy formula in liquid drop model is denoted by $E_s$ , determine the ratio of the surface energy per nucleon for ${}_{13}^{27}\text{Al}$ to ${}_{30}^{64}\text{Zn}$ ,	4	CO1
Q.3	Classify the different elementary particles based on their spins.	4	CO4
Q.4	List the similarities of nucleus with a liquid drop.	4	CO2
Q.5	Calculate the time required for 10% of a pure sample of ${}_{90}^{232}\text{Th}$ to disintegrate. $T_{1/2} = 1.4 \times 10^{10}$ years.	4	CO2
<b>SECTION B</b> <b>(4Qx10M= 40 Marks)</b>			
Q.6	Binding Energy of a light nuclei (Z, A) in MeV is given by the approximate formula.  $B(A, Z) \cong 16A - 20 A^{\frac{2}{3}} - \frac{3}{4} Z^2 A^{-\frac{1}{3}} + 30 \frac{(N - Z)^2}{A}$ N = A- Z is the neutron number, Determine the value of Z for most stable isobar for a given A. [Hint $\frac{\partial B}{\partial Z} = 0$ ]	10	CO2
Q.7	Explain briefly the tunnel theory of alpha decay	10	CO2
Q.8	Find the spin and Parity of the following nuclei based on Shell model.  ${}^3_2\text{He}$ , ${}^{20}_{10}\text{Ne}$ , ${}^{27}_{13}\text{Al}$ , ${}^{41}_{21}\text{Sc}$	10	CO1
Q.9	Differentiate between the Direct and Compound nuclear reaction mechanism with examples.	10	CO1
OR			

	Define attenuation and hence obtain expression for mass attenuation coefficient of gamma radiations.		<b>CO1</b>
<b>SECTION-C</b> <b>(2Qx20M=40 Marks)</b>			
Q.10	a) Describe the phenomena of Compton scattering. Obtain an expression for Compton shift.	<b>15</b>	<b>CO3</b>
	b) Explain the phenomena of pair annihilation.	<b>05</b>	
Q.11	Obtain an expression for threshold value for a nuclear reaction to occur. Assume the reaction to be of the type $a + A \rightarrow b + B$ .	<b>20</b>	<b>CO2</b>
	OR		
	a) Discuss the different mechanisms by which charge particles interact with matter.	<b>10</b>	<b>CO3</b>
	b) Determine the minimum energy that a photon should have for it to split an alpha particle into a tritium and a proton. (The masses of ${}^4_2\text{He}$ , ${}^3_1\text{H}$ and ${}^1_1\text{H}$ are 4.0026 amu, 3.0161 amu and 1.0073 amu respectively and $1 \text{ amu} \approx 938 \text{ MeV}$ .)	<b>10</b>	