

<b>Name:</b>	
<b>Enrolment No:</b>	

**UNIVERSITY OF PETROLEUM AND ENERGY STUDIES**  
**End Semester Examination (Online Mode)**

**Course: Safety, health and environment in petroleum industry**  
**Program: B. Tech APE GAS**  
**Course Code: ENVO 405**

**Semester: V**  
**Time : 03 hrs.**  
**Max. Marks: 100**

**Instructions:**

**SECTION A**

**Each Question will carry 5 Marks**

S. No.		Marks	CO
Q 1	Fire and explosion accidents can be prevented using the knowledge -----	5	CO5
Q2	What is the importance of occupational health? a) It concentrates relationship between work and individual health and prevent from ill health. b) It provides the legal obligation c) Occupational health is important to understand the hygiene requirement d) It is important to maintain good relationship between employee and employer	5	CO5
Q3	What hazard do safety goggles protect against? a) Heat b) Dust c) Strong light	5	CO1
Q4	What is impossible for gloves to protect against? a) Hazardous substances b) Cold or heat c) Rotating parts	5	CO2
Q5	Hazardous substances are classified by category. What are some of these categories?	5	CO3
Q6	What is a) A hazard b) A risk	5	CO4

**SECTION B**

**Each question will carry 10 marks**

Q 7	What are the impacts of drilling mud discharge in ocean environment?	10	CO1
Q8	What are the outcomes of Hazard identification and Operability (HAZOP) analysis?	10	CO2
Q9	Explain the Fire and Explosion characteristics of a material.	10	CO3

Q10	Explain Primary and Secondary Key words of HAZOP with examples	10	CO4
Q11	What are the stability classifications of atmosphere in dispersion modeling?	10	CO5

**SECTION-C**

**Each Question carries 20 Marks.**

Q 12	<p>Why Health safety and Environmental Management is important in Petroleum and offshore industry? Explain with the help of an industry disaster with its causes and consequences.</p> <p style="text-align: center;">Or</p> <p>Explain the steps to draw a flammability diagram. Draw a flammability diagram for Methane (CH<sub>4</sub>).</p> <p style="text-align: center;"><math>CH_4 + 2O_2 \rightarrow CO_2 + 2H_2O</math></p> <table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 40%;">Characteristics of Methane</td> <td></td> </tr> <tr> <td>Flammability limit in air</td> <td>LFL: 5.3% fuel in air</td> </tr> <tr> <td></td> <td>UFL: 15% fuel in air</td> </tr> <tr> <td>Flammability limit as pure form</td> <td>LFL: 5.1% fuel in oxygen</td> </tr> <tr> <td></td> <td>UFL: 61% fuel in oxygen</td> </tr> <tr> <td>LOC</td> <td>12% oxygen</td> </tr> </table>	Characteristics of Methane		Flammability limit in air	LFL: 5.3% fuel in air		UFL: 15% fuel in air	Flammability limit as pure form	LFL: 5.1% fuel in oxygen		UFL: 61% fuel in oxygen	LOC	12% oxygen	<b>20</b>	<b>CO5</b>
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