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UNIVERSITY OF PETROLEUM AND ENERGY STUDIES
End Semester Examination, Dec 2019

Program Name: B. Tech (SoCS)
IOT, Block chain, OGI, OSS, CSF, Big Data, AI+ML, Devops
Course Name : Physics
Course Code : PHYS 1008
No. of page/s: 02

Semester – I

Max. Marks :100
Duration : 3Hrs

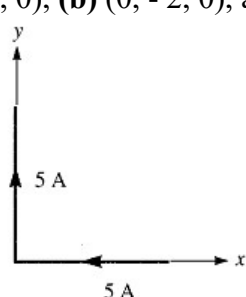
All questions are compulsory.
Question numbers to be written very clearly.
All bold representations are vectors.

SECTION A (All Questions are compulsory)

1	Discuss characteristic properties of a LASER beam. How is it different from ordinary light beam?	[4]	CO1
2	The numerical aperture of a fiber is 0.25 and the fractional refractive index is 0.02. Determine the refractive indices of the core and cladding of a fiber	[4]	CO1
3	What do you understand by a wave function? Discuss the properties of a well-behaved wave function.	[4]	CO4
4	Given the ket $ \psi\rangle = 3 0\rangle - 2i 1\rangle$ Find its normalized state.	[4]	CO5
5	Write the Maxwell's equations in differential and integral forms for static fields.	[4]	CO2

SECTION B

(All Questions are compulsory with an internal choice in Questions 10 and 11)

6	Discuss the working of He Ne LASER with the help of energy level diagram. Discuss all the processes involved.	[8]	CO1
7	Using Gauss's law, derive an expression for the electric flux density for an infinite sheet of charge.	[8]	CO2
8	An infinitely long conductor is bent into an L shape as shown in the figure below. If a direct current of 5 A flows in the conductor, find the magnetic field intensity at (a) (2, 2, 0), (b) (0, -2, 0), and (c) (0, 0, 2). Take the origin at the bend. 	[8]	CO3
9	What do you understand by phase velocity (v_p) and group velocity (v_g) of matter waves. Prove that	[8]	CO4

$$v_g = v_p - \lambda \frac{dv_p}{d\lambda}$$

10	What were the observations of photoelectric effect? Discuss how classical wave theory failed and how quantum theory of light was able to explain those observations. OR	[8]	CO4
11	State Heisenberg's uncertainty principle and based on it explain why electron cannot exist inside the nucleus.	[8]	CO4

SECTION C

(All Questions are compulsory with an internal choice in Questions 13 and 14)

12	(a) Using Ampere's law and continuity equation, obtain an expression for the displacement current density. (b) Differentiate between classical computing and quantum computing.	[10] [10]	CO3 CO5
13	(a) Derive Schrodinger's wave equation in time independent form. Write the expression for Hamiltonian. (b) Calculate the lowest energy of an electron confined in a 3-D cubical box of each side 1Å. (ii) Find the temperature at which the average energy of the molecules of a perfect gas would be equal to the lowest energy of the electron, $k_B = 1.38 \times 10^{-23}$ J/K.	[10] [10]	CO4 CO4
14	OR (a) What is Compton Effect? Derive an expression for the Compton shift in wavelength by considering inelastic scattering of a photon with an electron. (b) A metallic surface, when illuminated with light of wavelength λ_1 , emits electrons with energies upto a maximum value E_1 , and when illuminated with light of wavelength λ_2 , where $\lambda_2 < \lambda_1$, it emits electrons with energies upto a maximum value E_2 . Prove that plank's constant h and the work function ϕ of the metal are given by	[10] [10]	CO4 CO4

$$h = \frac{(E_2 - E_1)\lambda_1\lambda_2}{c(\lambda_1 - \lambda_2)} \quad \text{and} \quad \phi = \frac{E_2\lambda_2 - E_1\lambda_1}{(\lambda_1 - \lambda_2)}$$

Values of constants:

Constant	Standard Values
Planck's Constant (h)	6.63×10^{-34} Joule-sec
Permittivity of free space (ϵ_0)	8.854×10^{-12} Farad/meter
Velocity of Light c	3×10^8 m/sec
Boltzmann constant (k_B)	1.38×10^{-23} J K ⁻¹
Rest mass of an Electron	9.11×10^{-31} Kg
Charge of electron	1.6×10^{-19} C